Phytochemical Analysis and Standardization of Pedalium murex Linn. Extract through HPLC Methods

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ABSTRACT

Aim: The study aims screening of phytochemicals and standardization of extract of Pedalium murex Linn. fruits. Methods: The extraction was carried out by maceration technique using 70% ethanol and 30% water. The aqueous ethanolic extract obtained was estimated for total phenolic and flavonoid content against the standard gallic acid and quercetin. Liquid–liquid partitioning method using solvents n-hexane, chloroform, ethylacetate provided the different fractions of aqueous ethanolic extract. Standardization was done with highly sensitive, accurate, easy and cost effective high performance liquid chromatographic method. Results: The preliminary phytochemical screening of extractive organic substances of P. murex fruits revealed the presence of alkaloids, saponins, tannins, flavonoids, sugar, glycosides, phenols and sterols. The total phenolic content determined using linear regression equation (y = 0.0125x - 0.0096) \( R^2 = 0.9982 \) was found to be 27.1 ± 0.72 mg/g equivalent of gallic acid while the flavonoid content was expressed as 176 ± 0.79 mg/g equivalent of quercetin. The mother extract and the aqueous fraction revealed peaks complementary with the chromatogram of standard quercetin addressing presence of quercetin. Conclusion: The study helped in developing the phytochemical profile of the extract and the standardization performed in the study confirmed the presence of quercetin in the mother extract and fractions. Key words: Maceration, Phytochemical screening, Liquid-liquid partitioning, Phenolic content, Flavonoid content, HPLC.

INTRODUCTION

Various health ailments have been traditionally treated using medicinal plants since ancient times. Plants are the reservoir of naturally occurring bioactive compounds for drug discovery. These compounds serve as remedy for variety of diseases and are of extreme interest in modern therapeutics.[1] Pedalium murex L. commonly designated worldwide as large caltrops has been consistently evaluated, past five decades, for its biological activity and medicinal properties. The plant belonging to the family Pedaliaceae is also called as Gokshura, Gokantaka, Bara gokhru, Khaar-e-kkhasak, Peru neransi and with few more names in other languages. It is commonly found throughout India especially in Southern coastal region i.e. Kokan, Gujrath, Kathiawar, Ceylon, Peninsula and Deccan. The plant is also distributed over Pakistan, Sri Lanka, Mexico and Africa.[2] It is an annual, branched, succulent herb about 15-40 cm in height with spiny, four angled brownish colored fruits. Leaves are ovate oblong, simple, opposite, fleshy, glabrous with minute scales on lower sides.[3] The plant flowers from September to November and fruits are observed in period November to January. Ayurvedic formulations use gokshura fruits for treatment of urinary disorders.[4] Traditionally, the admixture produced from stem, fruits and/or leaves are effective as demulcent and diuretic. The plant is also used in urinary disorders like incontinence of urine, dysuria, gonorrhea and spermatorrhoea. It is believed to relieve strangury and dissolve calculi. Leaves are largely used to heal ulcers and as curry in splenic enlargement.[5][6] Decoction of roots is used as antibilious. So far, antimicrobial, anti-bacterial, antioxidant, anti-inflammatory, nephroprotector, antihyperlipidemic, antidiabetic, antiulcer and aphrodisiac activities have been studied on the plant.[7] The glycosides and glucuronides like dinatoin and diosmetin along with several flavonoids from leaves and flowers were isolated. The steroidal content from petroleum ether extract has found to possess fertility enhancing properties. Moreover, triacytanyl dotriacontanoate, 2′,4′,5′-trihydroxy 5,7-dimethoxy flavones, tetraatriacontanol octacosaanote and heptatriacontan-4-one from fruits have been isolated previously.[8] The fruits are found to be rich in soluble proteins, sapogenins and flavonoids.[1] The plant being loch of naturally occurring bioactive metabolites, the work was undertaken to analyze the phytochemical profile and develop fingerprinting of hydro alcoholic extract and fractions for standardization of extract.

METHODS

Plant material

The fruits were collected from local market and authenticated by Dr. N. Dongarwar, Professor, Department of Botany, R.T.M. Nagpur University, Nagpur. The specimen voucher number is 1078.

Extraction

The dried fruits were subjected to size reduction. The coarsely powdered drug (500 g) was macerated for 48 hr in aqueous Ethanolic solvent (70% ethanol: 30% water) with intermediate stirring. The obtained extract was filtered by using Whatman filter paper no. 42 and concentrated using a rotary vacuum evaporator to get the semisolid residue.[9] The percent yield was calculated as: Weight of extract / Weight of drug taken ×100
Phytochemical analysis
Battery of chemical tests was performed to ascertain presence of phytoconstituents. The hydroalcoholic extract was scrutinized for the presence of cardiac glycosides, alkaloids, flavonoids, steroids, glycosides, saponins, phenols and tannins according to standard procedures.[10]

Quantitative estimation
Estimation of total phenolic content
Folin Ciocalteau method was used for determination of phenolic content. Briefly, stock solution was prepared by dissolving 100 mg of extract, in 100 mL of methanol. This was further diluted 3 times and 1 mL of this resultant solution was exposed to 8 mL of methanol and 0.5 mL of 2N Folin Ciocalteau reagent (FCR), along with 1.5 mL of 20% sodium carbonate followed by a vigorous mixing. Gallic acid in different concentrations was used as a reference standard. Finally, the prepared solution was incubated in dark for 2 h and absorbance was recorded at 765 nm against blank using UV spectrometer. The total phenolic content was expressed in terms of mg/g gallic acid equivalent.

Estimation of total flavonoid content
The colorimetric method, was used for determination of flavonoid content. Extract (10 mg/mL) in methanol was transferred in test tube containing 20% aluminum trichloride, followed by addition of few drops of acetic acid and further diluted with methanol up to 5 mL. The absorption was taken at 415 nm after 40 min incubation in dark. Blank was prepared using same method except the incorporation of extract. The absorption of standard rutin solution (0.5 mg/mL) in methanol was measured under the same conditions which were used for extract. The amount of flavonoids in sample was calculated as mg/g rutin equivalent, using formula: 

\[ X = \frac{(A. \ m)}{(A_0. \ m)} \]

Where X is the flavonoid content, mg/g sample in RE, A is the absorbance of sample solution, Ao is the absorbance of standard rutin solution, m is the weight of sample in mg and m₁ is the weight of rutin in the solution in mg.[11]

Fractionation
Liquid-liquid partitioning method was used for fractionation. About 10 g extract was dissolved in 200 mL of water and sonicated three times for 25 -30 min. Further the solution was transferred in separating funnel and allowed to set for 10 min. n-Hexane 25 mL was placed in separating funnel containing extract, slowly tilted upside down for 100 times and allowed to set. Color in n-hexane layer indicated separation of compounds. The colored fraction was placed in preweighed petri plate. The procedure was repeated till n-hexane layer became transparent. The method was repeated for chloroform and ethyl acetate Figure 3. Eventually, the remaining water fraction was filtered with Whatman filter paper and evaporated till dryness to derive water fraction.[12]

HPLC
Chemicals
Standard quercetin was purchased from Sigma Mumbai. The solvents used were of HPLC grade.

Preparation of standard solution
Accurately weighed 10 mg of quercetin placed in 10 ml volumetric flask was dissolved in small portion of methanol and volume was made up to 10 ml (1000 ppm stock solution). Further, 1 ml of stock solution was diluted with 10 ml methanol, to get final concentration of 100 ppm.

Preparation of test solution
The test solutions of extract and fractions (100 ppm) were prepared using same method as that for standard.

Instrumentation
The HPLC chromatographic studies were performed by reversed phase HPLC UV method using an isocratic elution. The system composed of a LC-20AD pump, a SPD-M20A diode array detector (Shimadzu, Japan), a reverse phase PrincetonSPHER-100 C₁₈ column (250 mm x 4.6 mm, 5 µm particle size) operated by LC-Solution data acquisition software. The mobile phase used was acetonitrile and 2% v/v acetic acid (pH 2.60) (40%: 60% v/v) with flow rate 1.3 mL/min, 35°C column temperature and detection was done at 370 nm Table 4.

Procedure
The standard solution and prepared solution were injected one after the other using Hamilton syringe and chromatogram were recorded on set parameters.

RESULTS AND DISCUSSION
Biologically active compounds are present at very low concentration in plants. Proper extraction methods cause less damage to the functional properties of phytoconstituents and ease better yield.[13] The aqueous ethanolic maceration gives better yield with no damage to the thermo labile components. The Table 1 represents the percent yield of extracts and fractions. The percent yield is the ratio of the weight in grams of extract to the weight of powdered drug taken, multiplied by 100.[13] The preliminary phytochemical screening of aqueous ethanolic extract of P. murex fruits revealed the presence of alkaloids, saponins, tannins, flavonoids, sugar, glycosides, phenols and steroids[14] Table 2. Figure 1 and 2 shows the calibration curve of gallic acid and quercetin. The total phenolic content determined using linear regression equation (y =

<table>
<thead>
<tr>
<th>Sr.no.</th>
<th>Extract/ fraction</th>
<th>Percent yield</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>Aqueous ethanolic extract</td>
<td>2.8%</td>
</tr>
<tr>
<td>2</td>
<td>n-Hexane</td>
<td>0.76%</td>
</tr>
<tr>
<td>3</td>
<td>Chloroform</td>
<td>0.38%</td>
</tr>
<tr>
<td>4</td>
<td>Ethylacetate</td>
<td>0.43%</td>
</tr>
<tr>
<td>5</td>
<td>Aqueous</td>
<td>1.4%</td>
</tr>
</tbody>
</table>

Table 1: Percent yield.
Mahajan and Itankar. Standardization of Pedalium murex Linn. Extract

0.0123x - 0.0096 \( R^2 = 0.9962 \) represents correlation coefficient) was found to be 27.1 ± 0.72 mg/g equivalent of gallic acid while the flavonoid content was expressed as 17.6 ± 0.79 mg/g equivalent of quercetin (Table 3).\(^{15}\) Standardization of herbal drugs is a matter of interest and concern for herbal drug industry.\(^{16}\) The quality of drugs may get affected due to parameters like harvesting, storage, processing methods, adulterants as well as during preparation of formulation. Herbal drugs are unacceptable unless they are properly characterized and authenticated.\(^{17}\) With the emergence of new trends in chromatographic techniques it is possible to produce rapid, easy, accurate, cost effective and reproducible methods for standardization.\(^{18,19}\) High performance liquid chromatography (HPLC) is highly sensitive method used for quality assessment of herbal drugs, extracts, products and formulations.\(^{20,21}\) Flavonoids are the major secondary metabolites present in P. murex. Figure 4. Represents the chromatogram of standard drug quercetin with retention time 4.094 at 370 nm. The retention peaks were considered for standardization of extract and fractions. The whole aqueous ethanolic extract and fractions including n-hexane, chloroform and ethylacetate revealed presence of multiple peaks at variable retention time which confirms the presence

### Table 2: Phytochemical analysis of aqueous ethanolic extract.

<table>
<thead>
<tr>
<th>Sr.no</th>
<th>Phytoconstituents</th>
<th>Test 1</th>
<th>Test 2</th>
<th>Observation</th>
<th>Inference</th>
</tr>
</thead>
<tbody>
<tr>
<td>1.</td>
<td>Sterols</td>
<td>Salkowski</td>
<td>Lieberman burchard</td>
<td>+</td>
<td>Sterols present</td>
</tr>
<tr>
<td>2.</td>
<td>Alkaloids</td>
<td>Dragendorff’s reagent</td>
<td>Mayer’s reagent</td>
<td>+</td>
<td>Alkaloids present</td>
</tr>
<tr>
<td>3.</td>
<td>Saponins</td>
<td>Foam test</td>
<td>+</td>
<td>Saponins present</td>
<td></td>
</tr>
<tr>
<td>4.</td>
<td>Tannins</td>
<td>Ferric chloride test</td>
<td>Lead acetate test</td>
<td>+</td>
<td>Tannins present</td>
</tr>
<tr>
<td>5.</td>
<td>Flavonoids</td>
<td>Shinoda test</td>
<td>+</td>
<td>Flavonoids present</td>
<td></td>
</tr>
<tr>
<td>6.</td>
<td>Sugars</td>
<td>Molisch's test</td>
<td>Barfoed's test</td>
<td>+</td>
<td>Sugars present</td>
</tr>
<tr>
<td>7.</td>
<td>Proteins</td>
<td>Biuret test</td>
<td>Xanthoproteic test</td>
<td>-</td>
<td>Proteins absent</td>
</tr>
<tr>
<td>8.</td>
<td>Amino acid</td>
<td>Ninhydrin test</td>
<td>-</td>
<td>Amino acids absent</td>
<td></td>
</tr>
<tr>
<td>9.</td>
<td>Glycosides</td>
<td>Sodium hydroxide test</td>
<td>+</td>
<td>Glycosides present</td>
<td></td>
</tr>
<tr>
<td>10.</td>
<td>Phenols</td>
<td>Ferric chloride test</td>
<td>+</td>
<td>Phenols present</td>
<td></td>
</tr>
</tbody>
</table>

Values are mean of triplicate determination (\(n=3\)) ± standard deviation

### Table 3: Estimation of total phenolic, flavonoids content of \(P. murex\) fruit.\(^{12}\)

<table>
<thead>
<tr>
<th>Sr.no</th>
<th>Estimations</th>
<th>Hydroalcoholic extract of (P. murex)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1.</td>
<td>Phenolic content</td>
<td>27.1 ± 0.72 mg/g equivalent of gallic acid</td>
</tr>
<tr>
<td>2.</td>
<td>Flavonoids content</td>
<td>17.6 ± 0.79 mg/g equivalent of quercetin</td>
</tr>
</tbody>
</table>

Figure 2: Calibration curve for standard Quercetin.

Figure 3: Fractionation using liquid-liquid partitioning.
Figure 4: Chromatogram of Standard Quercetin.

Figure 5: Chromatogram of Mother extract.

Figure 6: Chromatogram of n-Hexane fraction.

Figure 7: Chromatogram of Chloroform fraction.

Figure 8: Chromatogram of Ethylacetate fraction.

Figure 9: Chromatogram of aqueous fraction.
of variable phytoconstituents Figures 5-8. The water fraction revealed peak at retention time 4.085, 4.432, which complement with the chromatogram of standard quercetin addressing presence of quercetin Figure 9.[22]

**CONCLUSION**

Phytochemical assessment of plant is an important parameter which gives the basic idea of the secondary metabolites present in the variable parts of plant. Therefore, it is requisite to perform the analysis according to the standard procedures mentioned. However, HPLC is the advent in the chromatographic investigations which simplifies the qualitative and quantitative studies of herbal extracts and allows detection of variable compounds in extracts. Thus, the present study provides the phytochemical profile of the *P. murex* extract and also confirmed the presence of quercetin in its different fractions along with the mother extract.

**ACKNOWLEDGEMENT**

The authors would like to acknowledge Department of Pharmaceutical Sciences, R.T.M. Nagpur University, Nagpur for providing the necessary facilities to carry out the study.

**CONFLICT OF INTEREST**

The authors declare no conflict of interest.

**ABBREVIATIONS**

- **R²**: Correlation coefficient; **mg/g**: Milligram/Gram; **w/w**: Weight by weight; **mg**: Milligram; **mL**: Milliliter; **N**: Normal; **nm**: Nanometer; **mg/mL**: Milligram/Milliliter; **Min**: Minute; **h**: Hour; **A**: Absorbance of sample solution; **mo**: Weight of Rutin in the solution in mg; **Ao**: Absorbance of standard Rutin solution; **m**: Weight of sample in mg; **HPLC**: High performance liquid chromatography; **ppm**: Parts per million; **UV**: Ultraviolet; **Mm**: Millimeter; **µm**: Microgram; **v/v**: Volume by volume; **mL/min**: Millilitre/minute; **C**: Celsius; **+**: Present; **-**: Absent; **µm/mL**: Microgram/milliliter; **CHCl³**: chloroform.

**REFERENCES**


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**Table 4: HPLC parameters for standardization.**

<table>
<thead>
<tr>
<th>Sr. no.</th>
<th>Variables</th>
<th>Conditions</th>
</tr>
</thead>
<tbody>
<tr>
<td>1.</td>
<td>Marker</td>
<td>Quercetin</td>
</tr>
<tr>
<td>2</td>
<td>Mobile Phase</td>
<td>Acetonitrile: Acetic acid (2% V/V) (40: 60) PH 2.6</td>
</tr>
<tr>
<td>3</td>
<td>Wavelength (nm) of Detection</td>
<td>370 nm</td>
</tr>
<tr>
<td>4</td>
<td>Flow rate</td>
<td>1.3 ml/min</td>
</tr>
<tr>
<td>5</td>
<td>Temperature</td>
<td>35°C</td>
</tr>
</tbody>
</table>

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*Mahajan and Itankar.: Standardization of *Pedalium murex* Linn. Extract*
The present work deals with the development of phytochemical profile and the standardization of crude extract of *Pedalium murex* fruits with the help of most effective, sensitive, easy and accurate high performance liquid chromatographic technique (HPLC). The work performed includes extraction, phytochemical analysis, estimation of total phenolic and flavonoid content followed by fractionation of crude extract with the series of solvent viz. n-hexane, chloroform, ethylacetate and water. The HPLC studies of mother extract and fractions were carried out. The preliminary phytochemical screening revealed the presence of variety of phytoconstituents. The total phenolic content was found to be 27.1 ± 0.72 mg/g equivalent of gallic acid while the flavonoid content was expressed as 17.6 ± 0.79 mg/g equivalent of quercetin. The mother extract and the aqueous fraction revealed peaks complementary with the chromatogram of standard quercetin addressing presence of quercetin. Standardization of herbal drugs being a matter of interest and concern for herbal drug industry, the method was found to be most reliable and reproducible.

**Summary**

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Mrs. Renuka K. Mahajan is a doctoral student at the Department of Pharmaceutical Sciences, Rashtrasant Tukadoji Maharaj Nagpur University, Nagpur, Maharashtra, from where she has completed her Bachelors and Master’s degree. Her doctoral research is focused on Development and Evaluation of Polyherbal Formulations for its Pharmacognostic and Pharmacological Potential Against Diabetic Nephropathy.

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